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==== DISCUSSIONS =

## On the Synthesis of 2-Acylindoles by the Darzens Condensation

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Radul et al. recently reported [1] on the N-alkylation of unsubstituted and sybstituted isatins with  $\alpha$ -bromo ketones in benzene and toluene in the presence of bases (pyridine, triethylamine,  $K_2CO_3$ ). These reactions led to formation of the corresponding N-acetonyl- and N-phenacylisatins which are used as starting compounds in the dihydroindoledione-indole rearrangement. A disadvantage of the proposed procedure is poor solubility of initial isatins in the above solvents. In addition, high reactivity of the carbonyl group in the  $\beta$ -position and ready generation of anions from bromo ketones by the action of bases favor the Darzens condensation, and separation of the products by preparative thin-layer chromatography is difficult because of their similar mobilities in different solvent systems.



R = H, Alk,; R' = Alk, Ar, Ht; M = K, Na.

It should be noted that the Darzens condensation, which readily occurs in DMF in the presence of  $K_2CO_3$  [2] or NaH, is the first stage in the synthesis of 2-acylindoles and that epoxy ketones I are nothing but a latent form of epoxy acid salts II; the latter are products of intermolecular condensation of *ortho*-aminobenzoylformic acids with halomethyl ketones, which does not occur under usual conditions. Structure I is hidden by the isatin ring; its opening in alkaline or superbasic medium [DMSO–water (9:1)–NaOH (5–10%)] leads to the explicit form, hydroxy acid III.

Thus, for the intermolecular condensation to occur it is necessary to attach the  $R'COCH_2$  fragment of halomethyl ketone to  $C^3$  of the isatin molecule before ring opening. Therefore, the Darzens condensation is an intramolecular version of this process.

Taking into account that the same 2-acylindoles IV are formed from the Darzens condensation products I and from 1-[2-oxoalkyl(aryl, heteryl)]-2,3-dihydroindole-2,3-diones in the dihydroindoledione–indole rearrangement [3], there is no need of suppressing the first of these via reduction of the solvent polarity; moreover, it can be regarded as a simple procedure for transformation of isatins into 3-(1,2-dioxoethyl)dihydroindol-2-ones V.\*

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